

The Atmosphere and its importance to the Earth

1. The atmosphere what is it?
 - a. A thin envelope of gases that surrounds the Earth
2. What does it do?
 - a. Controls the weather, hydrologic cycle, and the climate of the Earth.
 - b. Allows for life to exist on Earth
3. Ancient Atmosphere
 - a. Composition
 - i. Carbon Dioxide
 - ii. Sulfur Dioxide
 - iii. Water vapor
 - iv. Nitrogen
 - b. What is the source of the atmosphere?
 - c. Could not support life – no oxygen
 - d. Conditions changed 3.8 Ga (billion years before present)
 - i. Water could condense (precipitation occurred)
 - ii. Life was generated (amino acids - aquatic microorganisms)
 - iii. CO₂ dissolves in water and is used by MO – tied up in rocks
 - iv. Oxygen enters atmosphere
 - v. Nitrogen remains – does not react with other chemicals
4. Present Atmosphere
 - a. Nitrogen 78%
 - b. Oxygen 21%
 - c. Argon ~ 1%
 - d. All other gases > 0.04%
5. Layers of the Atmosphere
 - a. Temperature stratifies the atmosphere
 - b. Troposphere
 - i. Lowest layer – 0 to 15 km
 - ii. Contains 80% of the mass of the atmosphere.
 - iii. Layer where weather occurs.
 - iv. Temperature decreases with height.
 - c. Stratosphere
 - i. Occurs between 15 and 50 km above the surface.
 - ii. Temperature increases with height.
 - iii. Contains ozone (O₃) that shields the surface from ultraviolet (UV) radiation and result in the warming of the layer.
 - d. Mesosphere
 - i. Temperature decreases with altitude
 - ii. Fewer ozone molecules to retain heat
 - e. Thermosphere
 - i. Outermost layer
 - ii. Blocks cosmic radiation
 - iii. Radiation increases temperature > 1000°C
6. Energy Balance

7. Pressure
 - a. Air Pressure – the weight of the overlying air
 - i. Measured in atm or bars (1 bar = 0.986 atm)
 - ii. 1 atm is the weight of the atmosphere at sea level ~ 1,035 g/cm²
 - b. Mixture of gases in the atmosphere is not uniform
 - c. Air pressure decreases as one moves away from the Earth's surface
 - d. As the pressure increases the density of the air increases
8. Water Vapor
 - a. Water content in the atmosphere is called the: **humidity**
 - b. Composition varies as a function of temperature.
 - i. Warm air holds > 40 g H₂O/Kg air
 - ii. Cold air holds < 1 g H₂O/Kg air
 - c. Addition/removal of water to the atmosphere requires/generates energy.
 - d. Importance to climate – water vapor transfers heat from warmer to colder regions
 - e. Latent heat – heat released or absorbed when matter changes state.
9. Latent Heat Cycle
10. Energy and the Atmosphere: Pressure, Water Vapor, & Weather
 - a. What drives energy within the Atmosphere?
 - i. Uneven Heating of Earth Surface
 - ii. Solar energy is a function of both
 - (1) Location
 - (2) Time of year
 - iii. Warm air rises
 - (1) As air rises it enters a lower pressure area and expands
 - (2) Expansion cools the air – the cold air hold less water
 - (3) Condensation occurs – clouds form (latent heat)
11. Global Air Circulation
 - a. Air masses move from high pressure areas to low pressure areas
 - b. Air Circulation Around a Pressure Low
 - c. Coriolis Effect
12. Wave-cyclones & Mid-Latitude Storms – Low and High Pressure Systems
 - a. Low Pressure
 - i. Air moves up
 - ii. Weather cloudy
 - iii. Usually associated with a cold front
 - b. High Pressure
 - i. Air descending
 - ii. Little to no clouds – sunny
13. Frontal Systems
 - a. Cold Front
 - i. Air mass in cold behind the front.
 - ii. Advancement forces the warm air up – clouds form
 - b. Warm Front –
 - i. Air mass is warm behind the front
 - ii. Warm is relatively stable – move slower than cold front
 - iii. Stationary Front – front is moving very slowly

- iv. Occluded front – cold front has caught the warm front
- c. Thunderstorms
 - i. As warm humid air rises condensation occurs when the air mass temperature reaches the dew point– releasing energy.
 - ii. Precipitation falls; some of the precipitation is carried high into the clouds by warm up drafting air. This increases the cloud growth and the energy of the system.
- 14. The Greenhouse Effect
 - a. Concentrations of Greenhouse Gases – Modern Era
 - b. Link between CO₂ and Global Temperature?
 - c. CO₂ in the Atmosphere
 - d. Temperature during the Industrial Revolution
 - e. Carbon Cycle
- 15. Acid Rain
 - a. All rain is acidic
 - b. Caused by elevated concentrations of NO_x and SO_x
 - c. Industry is the source of NO_x and SO_x